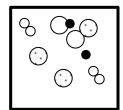
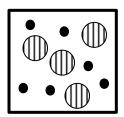
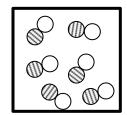
1) For the following 3 diagrams, please indicate whether each is a element, compound, or mixture. If the diagram is a mixture, please indicate the number of elements or compounds in the mixture.







- 2) Please state whether each of the following is a physical OR chemical change:
 - a) sodium reacts violently with water to produce hydrogen gas
 - b) gatorade crystals dissolve in water
 - c) your car burns gas
 - d) water boils on the stove top
- 3) List two elements that belong to each of the following groups in the periodic table:

a)	noble	gases:	
----	-------	--------	--

&	

b) alkali metals:

c) halogens:

	&
,	&

4) Draw the Bohr-Rutherford diagram for boron

5) Fill in the following table:

Symbol	Mass #	Atomic #	# of Protons	# of Electrons	# of Neutrons	Net Charge
F	19					0
K+	39		19			
O ²⁻	18			10		

6) Compare and contrast Ion & Isotope